

## Lesson Plan

**Name of the Faculty** : Sonia  
**Discipline** : Common for all Classes  
**Semester** : 2<sup>nd</sup>  
**Subject** : Applied Chemistry -II  
**Paper Code** : 170024  
**Lesson Plan Duration** : 15 weeks (from January 2018 to April 2018)

Week	Theory		Practical	
	Lecture Day	Topic (including assignment/ test)	Practical Day	Topic
1 <sup>st</sup>	1 <sup>st</sup>	General metallurgical terms and operations with reference to iron, copper and aluminium(Lecture-1)	1 <sup>st</sup>	Gravimetric analysis and apparatus used in gravimetric analysis
	2 <sup>nd</sup>	General metallurgical terms and operations with reference to iron(Lecture-2)		
	3 <sup>rd</sup>	General metallurgical terms and operations with reference to iron(Lecture-3)		
2 <sup>nd</sup>	1 <sup>st</sup>	Manufacture of steel- Open hearth process	1 <sup>st</sup>	Determination of percentage purity of commercial sample of blue vitriol using N/20 Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>
	2 <sup>nd</sup>	Alloys- definition and purpose of alloying		
	3 <sup>rd</sup>	Type of alloys – ferrous and non-ferrous alloys, properties and applications of ferrous alloys- invar, nichrome, stainless steel, alnico		
3 <sup>rd</sup>	1 <sup>st</sup>	Properties and applications of non-ferrous alloys – brass, bronze, duralumin, magnalium and solder	1 <sup>st</sup>	<b>Practice Practical 1 and 2</b>
	2 <sup>nd</sup>	<ul style="list-style-type: none"> <li>• <b>Home Work Checking</b></li> <li>• <b>Assignment-1 Distribution:</b> On any one of the three topics:               <ol style="list-style-type: none"> <li>i) List of iron, aluminium and copper metal ores and place of occurrences in India</li> <li>ii) Names of steel plants situated in India.</li> <li>iii) Enlist hydro power plants and nuclear power plants in India</li> </ol> </li> </ul>		
	3 <sup>rd</sup>	Definition of corrosion, its types and factors affecting corrosion rate		
4 <sup>th</sup>	1 <sup>st</sup>	Theories of dry (chemical) corrosion- Pilling Bedworth rule	1 <sup>st</sup>	Gravimetric estimation of moisture in the given coal sample (proximate analysis)
	2 <sup>nd</sup>	Theories of wet corrosion in acidic atmosphere by hydrogen evolution mechanism		
	3 <sup>rd</sup>	Definition of passivity in metals as per galvanic series		
5 <sup>th</sup>	1 <sup>st</sup>	Corrosion control: <ul style="list-style-type: none"> <li>• Metal coatings – Cathodic protection(Sacrificial protection and impressed current voltage), Cementation on Base Metal Steel – Application of Metal Zn (Sheradizing),Cr (Chromozing) and Al (Calorizing),</li> <li>• Inorganic coatings – Anodizing and</li> </ul>	1 <sup>st</sup>	Determination of percentage composition of volatile/non volatile matter in

		phosphating		the given coal sample
	2 <sup>nd</sup>	Corrosion control: <ul style="list-style-type: none"> <li>Organic coatings - use of paints varnishes and enamels</li> <li>Internal corrosion preventive measures-alloying (with reference to passivating, neutralizing and inhibition) and heat treatment (quenching, annealing)</li> </ul>		
	3 <sup>rd</sup>	<ul style="list-style-type: none"> <li><b>Class Work Checking/ Assignment-1 Checking</b></li> </ul>		
6 <sup>th</sup>	1 <sup>st</sup>	<ul style="list-style-type: none"> <li><b>Test-1</b></li> </ul>	1 <sup>st</sup>	Practice Practical 3 and 4
	2 <sup>nd</sup>	Definition of fuel, classification of fuels, characteristics of good fuel, relative merits of gaseous, liquid and solid fuels		
	3 <sup>rd</sup>	Calorific value-higher calorific value, lower calorific value		
7 <sup>th</sup>	1 <sup>st</sup>	Determination of calorific value of solid or liquid fuel using Bomb calorimeter and numerical examples	1 <sup>st</sup>	Gravimetric estimation of ash content in the given coal sample (proximate analysis)
	2 <sup>nd</sup>	Coal - types of coal and proximate analysis of coal <ul style="list-style-type: none"> <li><b>Assignment-2 Distribution</b></li> </ul>		
	3 <sup>rd</sup>	Fuel rating – Octane number and Cetane number, fuel-structural influence on Octane and Cetane numbers		
8 <sup>th</sup>	1 <sup>st</sup>	Gaseous fuels – chemical composition, calorific value and applications of natural gas (CNG), LPG, producer gas, water gas and biogas	1 <sup>st</sup>	Determination of viscosity of given liquid using Redwood viscometers
	2 <sup>nd</sup>	Elementary ideal on – hydrogen as future fuels, nuclear fuels		
	3 <sup>rd</sup>	<ul style="list-style-type: none"> <li><b>Revision</b></li> </ul>		
9 <sup>th</sup>	1 <sup>st</sup>	<ul style="list-style-type: none"> <li><b>Test-2</b></li> </ul>	1 <sup>st</sup>	Practice Practical 5 and 6
	2 <sup>nd</sup>	Definition of Lubricant and lubrication, type of lubrications –hydrodynamic, boundary lubrication with illustrative diagrams		
	3 <sup>rd</sup>	Classification of lubricants–liquid lubricants, solid lubricants with examples		
10 <sup>th</sup>	1 <sup>st</sup>	Classification of lubricants- semi-solid lubricants and synthetic lubricants with examples	1 <sup>st</sup>	Determination of flash point of given lubricating oil using Able’s flash point apparatus
	2 <sup>nd</sup>	Physical properties of lubricant –viscosity and viscosity index, cloud point and pour point, flash point and fire point, oiliness		
	3 <sup>rd</sup>	Chemical properties of lubricant - total acid value or number (TAV or TAN), carbon residue, emulsification factor and iodine value		
11 <sup>th</sup>	1 <sup>st</sup>	Designation of lubricating oils according to Society of Automotive Engineers (SAE)	1 <sup>st</sup>	Practice Practical 7
	2 <sup>nd</sup>	Cutting fluids – applications of cutting fluids, types and the factors that govern the selection of cutting fluids		
	3 <sup>rd</sup>	<ul style="list-style-type: none"> <li><b>Class Work Checking/ Assignment Checking-2</b></li> </ul>		
12 <sup>th</sup>	1 <sup>st</sup>	Definition and types with suitable examples and applications of- Ceramics, Refractory and Composite materials	1 <sup>st</sup>	To study the effect of metal

	<b>2<sup>nd</sup></b>	Glass-chemical composition and application of Soda <ul style="list-style-type: none"> <li>• <b>Assignment-3 Distribution</b></li> </ul>		coupling on corrosion of iron
	<b>3<sup>rd</sup></b>	Chemical composition and application of Borosilicate and lead glasses		
<b>13<sup>th</sup></b>	<b>1<sup>st</sup></b>	Paint, varnish and enamels- definition, constituents	<b>1<sup>st</sup></b>	Detection of iron metal in the given solution of rust(solution of rust in HCl be provided)
	<b>2<sup>nd</sup></b>	Advantages of Paint, varnish and enamels		
	<b>3<sup>rd</sup></b>	Definition of polymer, monomer and degree of polymerization		
<b>14<sup>th</sup></b>	<b>1<sup>st</sup></b>	Brief introduction to addition and condensation polymers with suitable examples (PE, PS, PVC, Teflon, Nylon -66 and Bakelite)	<b>1<sup>st</sup></b>	<b>Practice Practical 8 and 9</b>
	<b>2<sup>nd</sup></b>	Definition of plastics, thermo plastics and thermo setting plastics with suitable examples, distinctions between thermo plastics and thermo settings		
	<b>3<sup>rd</sup></b>	Applications of polymers in industry and daily life		
<b>15<sup>th</sup></b>	<b>1<sup>st</sup></b>	<ul style="list-style-type: none"> <li>• <b>Revision/ Home Work Checking</b></li> </ul>	<b>1<sup>st</sup></b>	<b>Practical Test</b>
	<b>2<sup>nd</sup></b>	<ul style="list-style-type: none"> <li>• <b>Class Work Checking/ Assignment-3 Checking</b></li> </ul>		
	<b>3<sup>rd</sup></b>	<ul style="list-style-type: none"> <li>• <b>Test-3</b></li> </ul>		